

## From Last Time:

Science of Cooking – BCBT100  
Fall 2016 – Bodwin/Tigges

## Milk

- WHAT IS MILK?** U.S. Code of Federal Regulations, Title 21, Vol. 8, Chpt 1, Pt 1240, subpart A, Section 1240.3(j), Release 13
- “Milk is the lacteal secretion, practically free from colostrum, obtained by the complete milking of one or more healthy cows. Milk that is in final package form for beverage use shall have been pasteurized or ultrapasteurized, and shall contain not less than 8 1/4 percent milk solids not fat and not less than 3 1/4 percent milkfat. Milk may have been adjusted by separating part of the milkfat therefrom, or by adding thereto cream, concentrated milk, dry whole milk, skim milk, concentrated skim milk, or nonfat dry milk. Milk may be homogenized.”

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## Milk

Protein-rich water with an emulsion of protein-coated fat globules

Water phase (aqueous):  
*Slightly acidic water (pH ~6.6)*  
*Protein bundles*  
*Lactose*

Fat phase:  
*Droplets of oil with a protein shell*

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## Mixtures

**Homogeneous**  
*Pure substances*  
*Solutions*

**Heterogeneous**  
*Bulk mixtures, melange*  
*Suspension/colloid, emulsion*  
**Emulsifiers and amphiphiles**

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## Milk Composition

1	Water	86.6 %
2	Fat	4.4 %
3	Lactose	4.6 %
4	Casein	2.7 %
5	Whey protein	0.6 %
6	NPN	0.1 %
7	Minerals	0.7 %
8	Salts	0.17 %
9	Vitamin and enzymes	0.13 %

NPN – Non-protein nitrogenous compounds

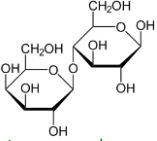
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## Sources of milk:

Species	Water	Fat	Casein	Whey	Lactose
Human	87.1	4.6	0.4	0.7	6.8
Cow	87.3	4.4	2.8	0.6	4.6
Buffalo	82.2	7.8	3.2	0.6	4.9
Goat	86.7	4.5	2.6	0.6	4.4
Sheep	82.0	7.6	3.9	0.7	4.8
Horse	88.8	1.6	1.3	1.2	6.2
Rat	79.0	10.3	6.4	2.0	2.6
Donkey	88.3	1.5	1.0	1.0	7.4
Reindeer	66.7	18.0	8.6	1.5	2.8
Camel	86.5	4.0	2.7	0.9	5.4

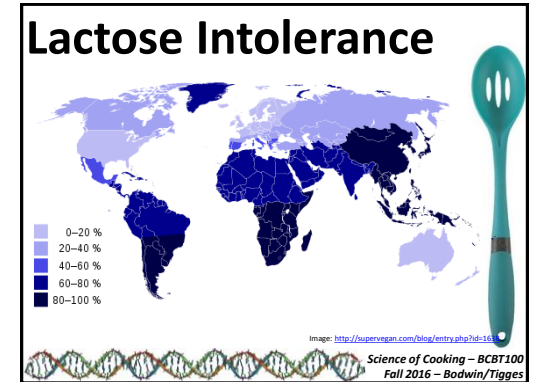
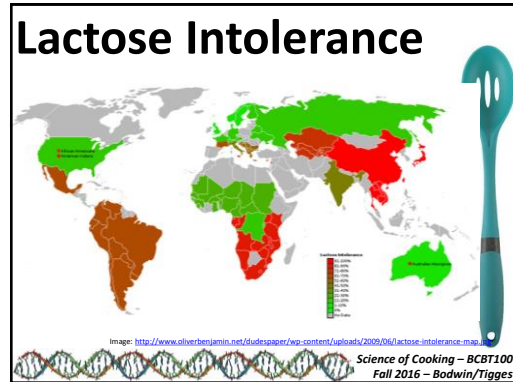
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## Lactose

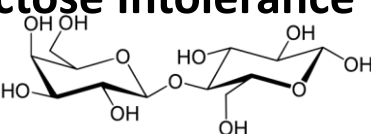


- Disaccharide - glucose and galactose prepared as separate molecules and condensed into "milk sugar" through the secretory cells
- Ability to digest (metabolize or "break down") lactose requires a special enzyme - lactase
- Lactase is produced in gut by children but levels decrease in adults.
- Northern Europeans maintain levels but only 30% of others can produce significant quantities

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## Lactose Intolerance



Lactase - hydrolytic enzyme  
Lactose passes through to gut  
Draws water in (osmosis)  
Bacterial digestion -  $CH_4(g)$ ,  $CO_2(g)$   
Cramps, gas, diarrhea

Image: <http://en.wikipedia.org/wiki/File:Beta-D-Lactose.png>

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## Purpose of Lactose

Glucose  
*Protected as disaccharide*  
Energy source

Galactose  
*Neural tissue*  
Make brains...

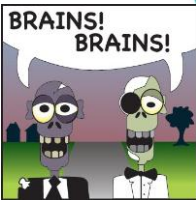


Image: [http://bloggy.blah-blah.blogspot.com/2012\\_01\\_01\\_archive.html](http://bloggy.blah-blah.blogspot.com/2012_01_01_archive.html)

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## Milk Protein - Casein

- Major single protein produced in most milk
- Key characteristics of casein
- Heat stable - well folded protein
- "floats" in micelle form (globs of protein arranged to keep the protein in solution)
  - Hydrophobic portion of protein in middle
- Calcium binds tightly to this protein - helps to carry calcium into the blood system!
- Four main forms of Casein - one "caps" micelles limiting the size
- At pH levels above 4.5, proteins are negatively charged and repel.
- When acid increases to pH lower than 4, proteins denature and are not charged - thus they bind to each other and "curdle"
- Body builders sometimes use this as a "slow-digesting protein" (why)

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# Casein Micelles

## kappa-Casein coating Calcium-binding

Image: <http://www.foodsci.uoguelph.ca/teicon/casein.gif>  
 Image: [http://openstaxware.org/images/thumb/9/32/AM\\_Micelle.jpg/300px-AM\\_Micelle.jpg](http://openstaxware.org/images/thumb/9/32/AM_Micelle.jpg/300px-AM_Micelle.jpg)

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# Acids and Bases

Acids = increase concentration of hydrogen ion (H<sup>+</sup>) when dissolved in water

Bases = decrease concentration of hydrogen ion (H<sup>+</sup>) when dissolved in water (increase OH<sup>-</sup> concentration)

H<sup>+</sup>(aq) + OH<sup>-</sup>(aq) → H<sub>2</sub>O(l)

“Neutralization”  
pH scale

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# Acids and Bases

Acids taste sour

Bases are slippery to the skin

- Muscle acid – lactic acid
- Vinegar – acetic acid
- Fruit acid – citric acid
- Oxalic acid – used in candy with citric acid

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# Milk Protein - Whey

- Soluble in acidic aqueous phase
- Many whey proteins are immunoglobins (antibodies for the young animal)
- Lactoglobulin has several sulfur atoms – provides flavor and odor to cooked milk
- Proteins in whey are used for animals as source of nutrition
- Under more extreme conditions than casein, whey proteins can form small clots – ricotta cheese
- These proteins help make ice cream... creamy

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# Milk Fat

Globules of fat in a phospholipid and protein shell (Emulsifiers)

Homogenization

Heat-stable globules

Cold breaks fat globules – ice, ice, baby

Fat soluble vitamins – A, D, E, K

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# Variations in Milk

Breed	Fat %	Casein %	Whey %	Lactose %
Brown Swiss	3.8	2.63	0.55	0.72
Holstein	3.56	2.49	0.53	0.73
Jersey	4.97	3.02	0.69	0.77

Delaval.com

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## Homogenization

Increase surface area  
Casein proteins coat – Negative

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## Sphere Math

Volume =  $\frac{4}{3} \pi r^3$   
Surface area =  $4 \pi r^2$

1 sphere, 2cm radius  
Volume =  $\frac{4}{3} \pi (2\text{cm})^3 = 34\text{cm}^3$   
Surface =  $4 \pi (2\text{cm})^2 = 50.\text{cm}^2$

Break into 2 spheres:  
Volume of each =  $17\text{cm}^3 = \frac{4}{3} \pi (x)^3 \rightarrow x = 1.6\text{cm}$   
Surface of each =  $4 \pi (1.6\text{cm})^2 = 32\text{cm}^2$   
Total surface =  $64\text{cm}^2 \rightarrow$  too much!

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