

Descriptions of Scale

Macroscale vs. Microscale

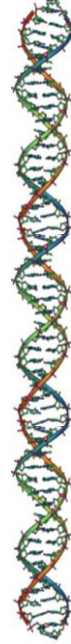
Chemistry & Molecular Biology
bridge these worlds

<http://htwins.net/scale2/>



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What is food made of?

Organic vs. Inorganic

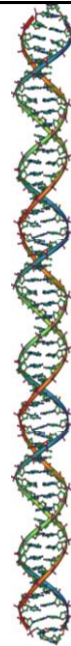
Organic = “from life”, contains C-H
bonds

Inorganic = no C-H bonds



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What is food made of?

Water

Inorganic components

Salts, minerals

“Small” Organic Molecules

Vitamins, metabolites

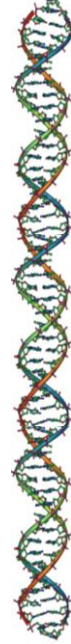
Macromolecules

Lipids, proteins, carbohydrates



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Water

H₂O!

Very small, simple

Essential to all life on Earth

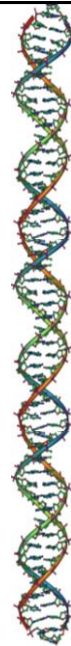
Search for Extraterrestrial Life

Most food is mostly water



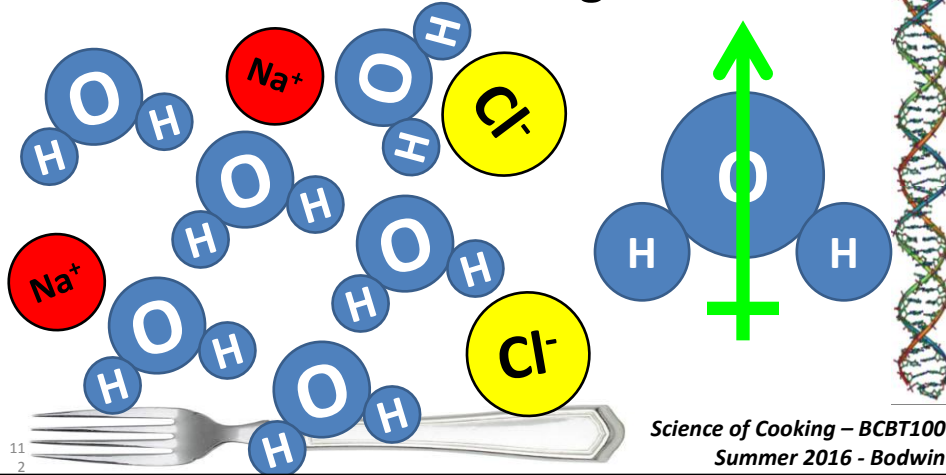
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Why is water liquid?

Water molecules are bent → polar
Polar molecules stick together



Water Content of Foods

Food	Water Content (%)
Meat	
Pork, raw, composite of lean cuts	53-60
Beef, raw, retail cuts	50-70
Chicken, raw meat without skin	74
Fish, muscle proteins	65-81
Fruits	
Berries, cherries, pears	80-85
Apples, peaches, oranges, grapefruit	85-90
Rhubarb, strawberries, tomatoes	90-95
Vegetables	
Peas (green)	74-80
Beets, broccoli, carrots, potatoes	80-90
Asparagus, beans, cabbage, cauliflower, lettuce	90-95

Source: <http://class.fst.ohio-state.edu/fst605/605%20pdf/Water.pdf>

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Water in Foods

Water content of some foods

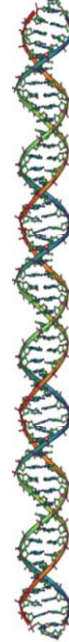
Food	Water content (%)
Beef	50 to 70
Chicken meat	74
Fish	65 to 81
Pears	80 to 85
Apples, peaches, oranges	85 to 90
Tomatoes, strawberries	90 to 95
Avocado, banana	74 to 80
Carrot, potato	80 to 90
Lettuce, lentils	90 to 95
Honey	20
Jam	28
Flour, rice	12
Milk powder	4

Source: <http://www.azaquar.com/en/doc/water-in-food>



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Inorganic Components

“Salts” – charged particles

Sodium chloride → Na⁺ and Cl⁻

Other trace minerals

Iron, potassium, calcium, magnesium, etc



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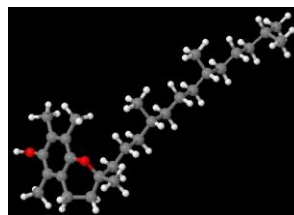
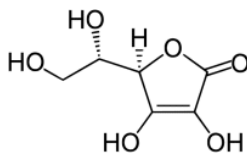
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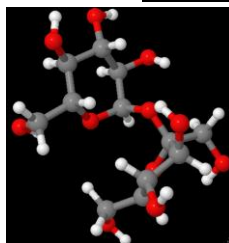
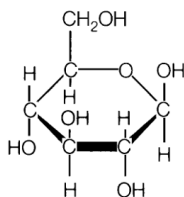
“Small” Organic Molecules

“Organic” = containing C and H

Vitamins



Sugars



Others



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BIG Food Molecules

Lipids

Proteins

Carbohydrates

DNA/RNA



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Types of Fats

Saturated vs. Unsaturated

Mono- vs Polyunsaturated

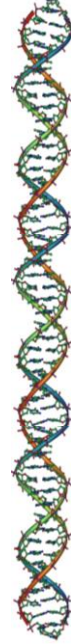
“Hydrogenated”

“Omega-3”



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Macromolecules

Polymers –

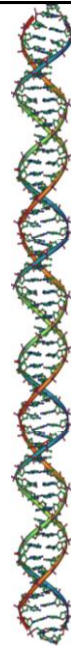
poly=“many”, meros=“parts”

Different “parts” result in
different function/properties

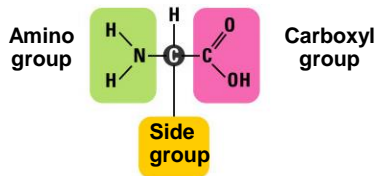


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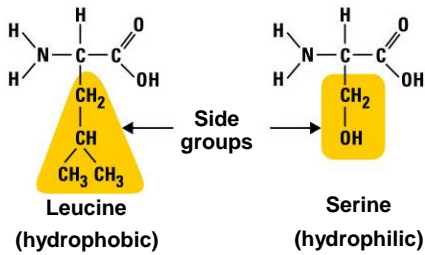
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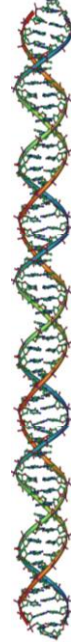
Amino Acids



Central Carbon
Carboxyl group
Amino group
Side chain - varies

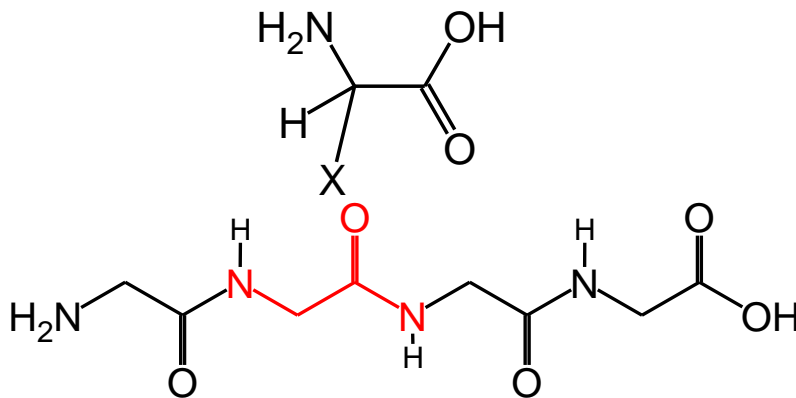
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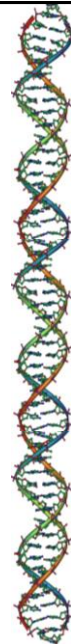


Proteins

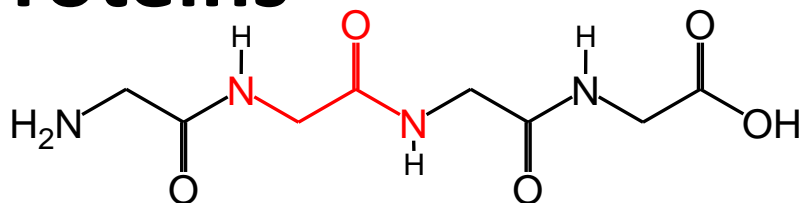
Polymers made of amino acids

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Proteins



Shape depends upon properties of
side chains interacting with water

Shape = Function

20 “letters”, many “words”

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4 Levels of Protein Structure

Primary – aa order

Secondary – near aa interactions

Tertiary – long range in 1 protein

Quaternary – clusters of proteins

Denaturing disturbs structure

Protein structure: http://en.wikipedia.org/wiki/File:Main_protein_structure_levels_en.svg

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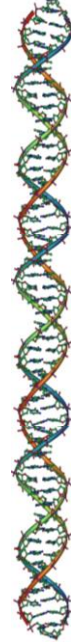
Carbohydrates

“Carbo” → carbon

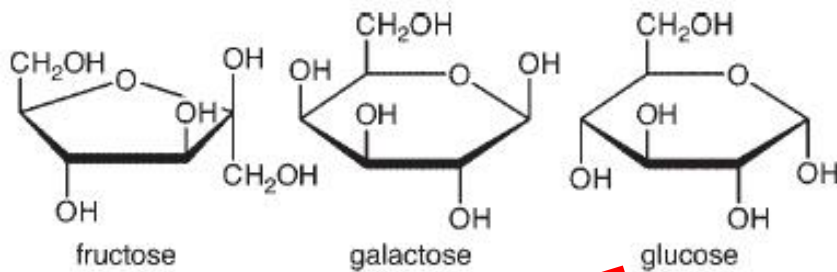
“hydrate” → water, O and H

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Common Sugars



Sucrose

Maltose

Lactose

Image: <http://en.wikipedia.org/wiki/File:Glycolysis2.svg>

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Common Sugars

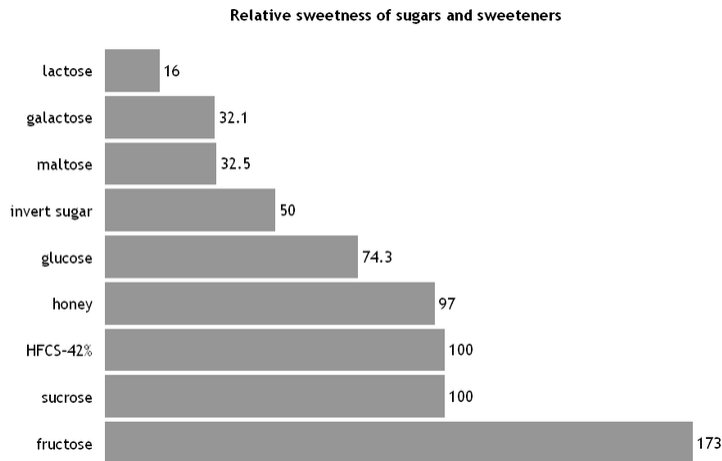


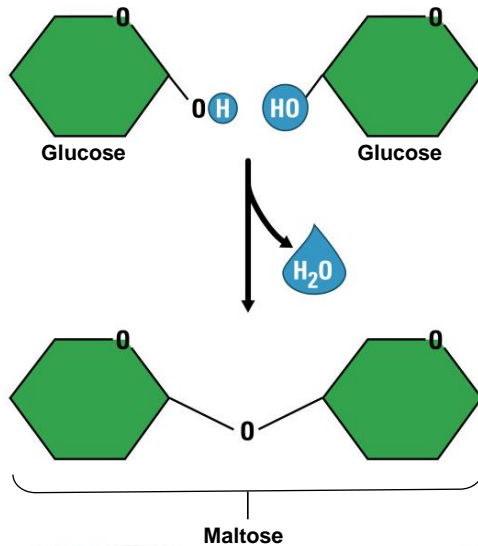
Image: <http://en.wikipedia.org/wiki/File:Relativesweetness.png>

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Disaccharides



Monosaccharides
react to form
disaccharides

Liberate water
Dehydration
Condensation

Reversible
Hydrolysis
"-ase" enzymes

Image: <http://en.wikipedia.org/wiki/File:Glycolysis2.svg>

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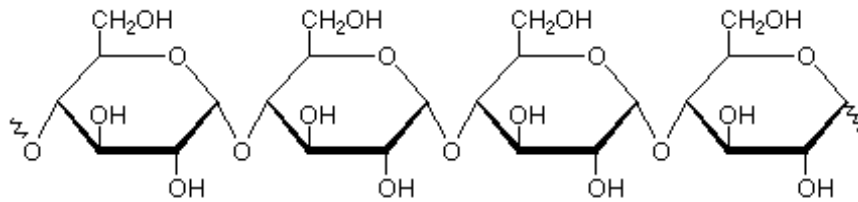
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Polysaccharides

Storage and structure

Starch, Glycogen, Cellulose

Sugar polymers



polysaccharide (amylose starch)

Image: <http://en.wikipedia.org/wiki/File:Amylose2.svg>



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Starch

Glucose polymers

Energy storage in plants

Potatoes, rice, grains

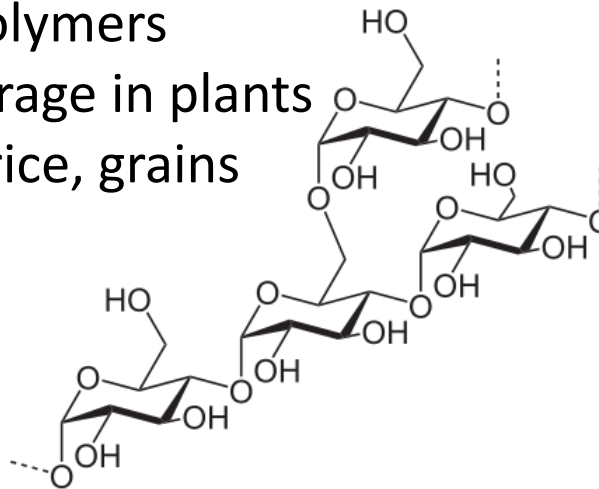


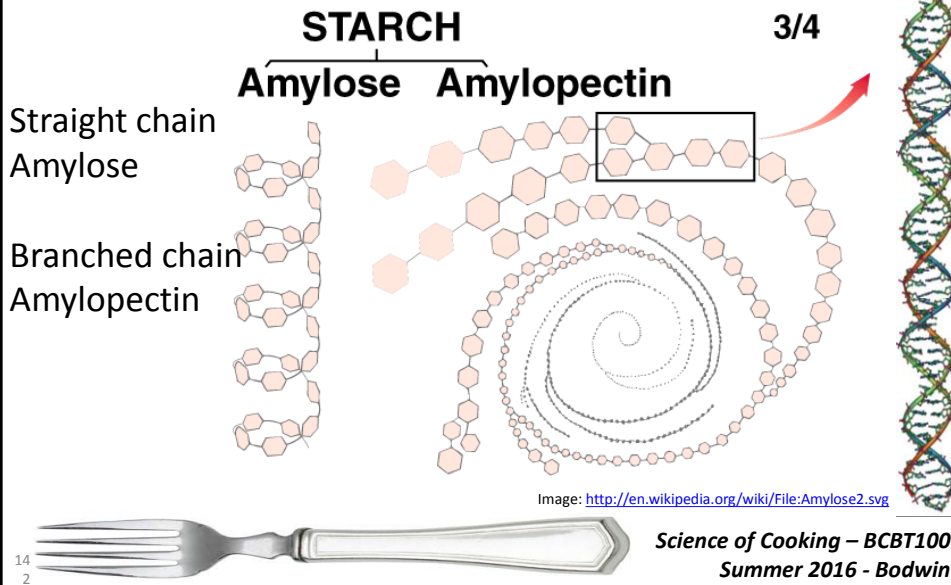
Image: http://en.wikipedia.org/wiki/File:Amylopektin_Sessel.svg



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Starch - Structure



Starch – In foods

Thickener – binds a LOT of water

Provides energy - amylase

Industrially:

Dextrose = glucose derived from
hydrolyzed starch

HFCS – dextrose treated with glucose
isomerase

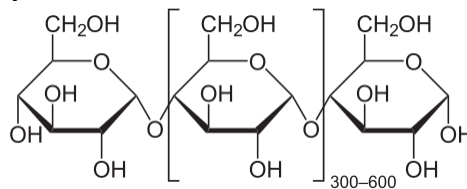


Image: <http://en.wikipedia.org/wiki/File:Amylose2.svg>

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Glycogen – “animal starch”

Highly branched glucose polymer

Energy storage

GLYCOGEN

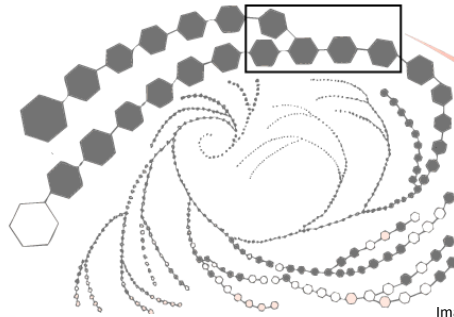


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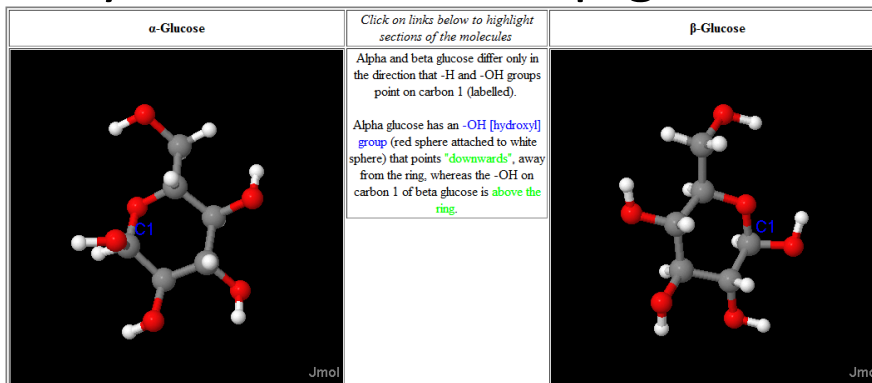
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Cellulose

Polymers made from β -glucose



Side-by-side animations:

<http://www.biopics.co.uk/JmolApplet/alphabetaajglucose2.html>

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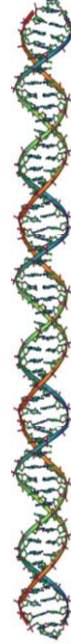
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Cellulose

Enzymes that break amylose
can't break cellulose

Rigid, tough *fibers* that make
plant cell walls and stalks

Cross-linking



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Cellulose - Dietary

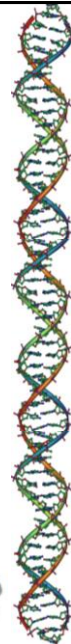
Insoluble Fiber

Highly modified cellulose, up to ~1/2 the
mass of a plant

Binds water, "feel full"

Draws water into gut

Fruits, vegetables, whole grains



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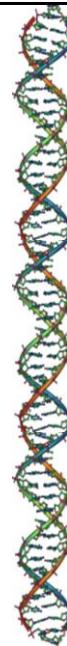
Cellulose - Dietary

Soluble Fiber

Highly modified cellulose

Forms gel with high water content

Water-soluble substances absorbed by gel – “intestine sweeper”

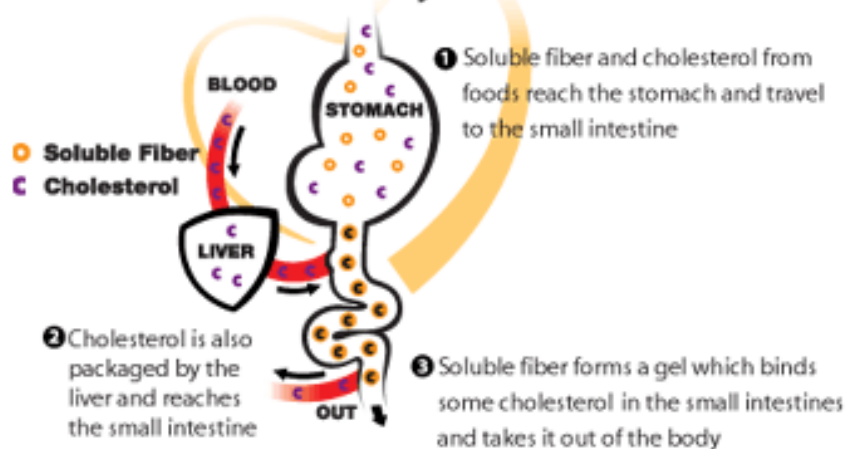


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Cellulose - Dietary

How Soluble Fiber May Lower Cholesterol



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Cellulose – Food source?

Cellulosic fiber is indigestible

Most animals lack enzymes to break down cellulose

Ruminants have bacteria in the gut that {partially} digest cellulose to glucose



Image: <http://www.publicdomainpictures.net/view-image.php?image=627&picture=black-cow>, <http://www.cvm.ncsu.edu/vhc/efac/rhm/>

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Interactions

Fats and water

Amphiphiles

Micelles

Emulsifiers

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