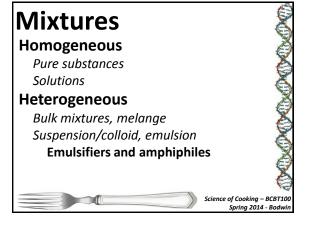
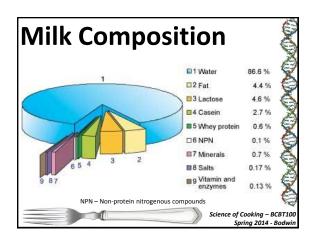
WHAT IS MILK? U.S. Code of Federal Regulations, Title 21, Vol. 8, Chpt 1, Pt 1240, subpart A, Section 1240.3(j), Release 13 "the lacteal secretion obtained from one or more healthy milk-producing animals, e.g. cows, goats, sheep, and water buffalo, including, but not limited to, the following: lowfat milk, skim milk, cream, half and half, dry milk, nonfat dry milk, dry cream, condensed or concentrated milk products, cultured or acidified milk or milk products..."

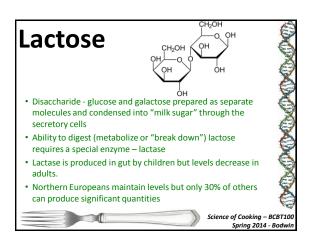
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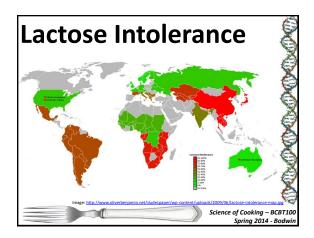
Protein-rich water with an emulsion of protein-coated fat globules Water phase (aqueous): Slightly acidic water (pH ~6.6) Protein bundles Lactose Fat phase: Droplets of oil with a protein shell

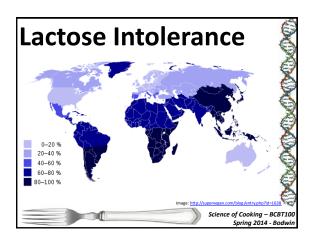


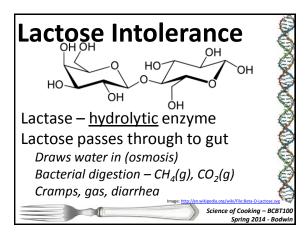


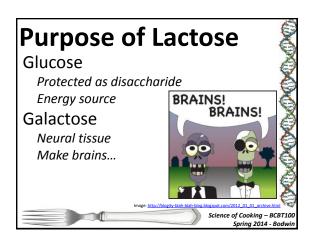
Sources of milk:						
Species	Water	Fat	Casein	Whey	Lactose	3
Human	87.1	4.6	0.4	0.7	6.8	Ę
Cow	87.3	4.4	2.8	0.6	4.6	P.
Buffalo	82.2	7.8	3.2	0.6	4.9	6
Goat	86.7	4.5	2.6	0.6	4.4	7
Sheep	82.0	7.6	3.9	0.7	4.8	Ş
Horse	88.8	1.6	1.3	1.2	6.2	6
Rat	79.0	10.3	6.4	2.0	2.6	6
Donkey	88.3	1.5	1.0	1.0	7.4	7
Reindeer	66.7	18.0	8.6	1.5	2.8	Ş
Camel	86.5	4.0	2.7	0.9	5.4	1
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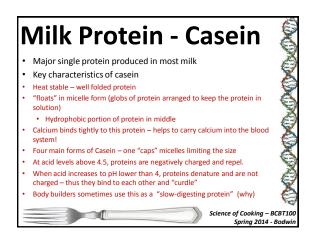


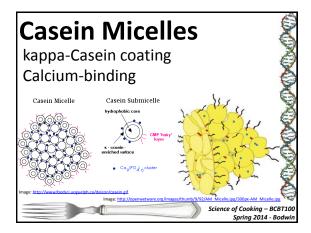












Milk Protein - Whey

- · Soluble in acidic aqueous phase
- Many whey proteins are immunoglobins (antibodies for the young animal)
- Lactoglobin has several sulfur atoms provides flavor and odor to cooked milk
- Proteins in whey are used for animals as source of nutrition
- · Under more extreme conditions than casein, whey proteins can form small clots - ricotta cheese
- · These proteins help make ice cream... creamy



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Acids and Bases Acids = increase concentration of hydrogen ion (H⁺) when dissolved in water Bases = decrease concentration of hydrogen ion (H⁺) when dissolved in water (increase OH- concentration) $H^+(aq) + OH^-(aq) \rightarrow H_2O(I)$ "Neutralization" Amino acid group pH scale Science of Cooking – BCBT100 Spring 2014 - Bodwin

