

Descriptions of Scale

Macroscale vs. Microscale

Chemistry & Molecular Biology
bridge these worlds

<http://www.scaleoftheuniverse.com/>



Science of Cooking – BCBT100
Spring 2014 - Bodwin



What is food made of?

Organic vs. Inorganic

Organic = “from life”, contains C-H
bonds

Inorganic = no C-H bonds



Science of Cooking – BCBT100
Spring 2014 - Bodwin



What is food made of?

Water

Inorganic components

Salts, minerals

“Small” Organic Molecules

Vitamins, metabolites

Macromolecules

Lipids, proteins, carbohydrates



Science of Cooking – BCBT100
Spring 2014 - Bodwin



Water

H₂O!

Very small, simple

Essential to all life on Earth

Search for Extraterrestrial Life

Most food is mostly water



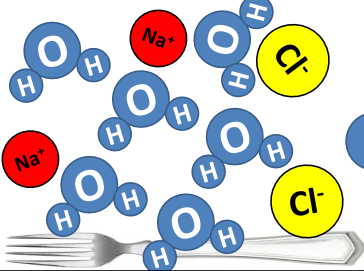
Science of Cooking – BCBT100
Spring 2014 - Bodwin



Why is water liquid?

Water molecules are bent → polar

Polar molecules stick together



Science of Cooking – BCBT100
Spring 2014 - Bodwin



Water Content of Foods

Food	Water Content (%)
Meat	
Pork, raw, composite of lean cuts	53-60
Beef, raw, retail cuts	50-70
Chicken, raw meat without skin	74
Fish, muscle proteins	65-81
Fruits	
Berries, cherries, pears	80-85
Apples, peaches, oranges, grapefruit	85-90
Rhubarb, strawberries, tomatoes	90-95
Vegetables	
Peas (green)	74-80
Beets, broccoli, carrots, potatoes	80-90
Asparagus, beans, cabbage, cauliflower, lettuce	90-95

Source: <http://class.fst.ohio-state.edu/fst605/605%20pdf/Water.pdf>



Science of Cooking – BCBT100
Spring 2014 - Bodwin



Water in Foods

Water content of some foods

Food	Water content (%)
Beef	50 to 70
Chicken meat	74
Fish	65 to 81
Pears	80 to 85
Apples, peaches, oranges	85 to 90
Tomatoes, strawberries	90 to 95
Avocado, banana	74 to 80
Carrot, potato	80 to 90
Lettuce, lentils	90 to 95
Honey	20
Jam	28
Flour, rice	12
Milk powder	4

Source: <http://www.azaquar.com/en/doc/water-in-food>



Science of Cooking – BCBT100
Spring 2014 - Bodwin

Inorganic Components

“Salts” – charged particles

Sodium chloride → Na^+ and Cl^-

Other trace minerals

Iron, potassium, calcium, magnesium, etc

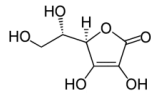


Science of Cooking – BCBT100
Spring 2014 - Bodwin

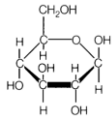
“Small” Organic Molecules

“Organic” = containing C and H

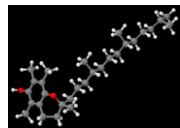
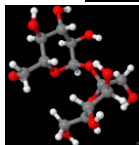
Vitamins



Sugars



Others



Science of Cooking – BCBT100
Spring 2014 - Bodwin

BIG Food Molecules

Lipids

Proteins

Carbohydrates

DNA/RNA



Science of Cooking – BCBT100
Spring 2014 - Bodwin



Lipids

Fats

Long chains of (mostly) C and H

Lipids are non-polar

Don't mix w/water = "hydrophobic"

"like dissolves like"

Molecule animations: <http://www.biotopics.co.uk/JmolApplet/contentstable.html>

Fatty acids Jmol:

http://www.mpcfaculty.net/mark_bishop/Bishop_Jmol_fatty_acids_triglyceride.htm



Science of Cooking – BCBT100
Spring 2014 - Bodwin



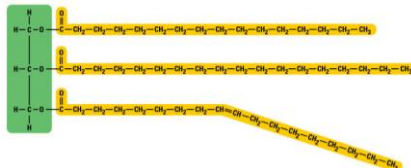
Fatty Acids/Triglycerides

Vinegar = 2 carbons

Water soluble

Stearic acid = 18 carbons

NOT water soluble



Science of Cooking – BCBT100
Spring 2014 - Bodwin



Types of Fats

Saturated vs. Unsaturated

Mono- vs Polyunsaturated

"Hydrogenated"

"Omega-3"



Science of Cooking – BCBT100
Spring 2014 - Bodwin

Macromolecules

Polymers –

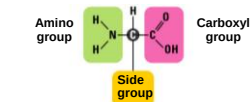
poly="many", meros="parts"

Different "parts" result in
different function/properties

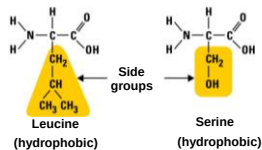


Science of Cooking – BCBT100
Spring 2014 - Bodwin

Amino Acids



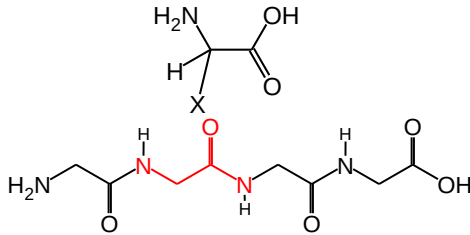
Central Carbon
Carboxyl group
Amino group
Side chain - varies



Science of Cooking – BCBT100
Spring 2014 - Bodwin

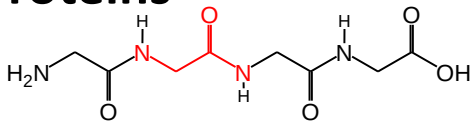
Proteins

Polymers made of amino acids



Science of Cooking – BCBT100
Spring 2014 - Bodwin

Proteins



Shape depends upon properties of
side chains interacting with water

Shape = Function

20 “letters”, many “words”



Science of Cooking – BCBT100
Spring 2014 - Bodwin

4 Levels of Protein Structure

Primary – aa order

Secondary – near aa interactions

Tertiary – long range in 1 protein

Quaternary – clusters of proteins

Denaturing disturbs structure

Protein structure: http://en.wikipedia.org/wiki/File:Main_protein_structure_levels_en.svg



Science of Cooking – BCBT100
Spring 2014 - Bodwin

Carbohydrates

"Carbo" → carbon

"hydrate" → water, O and H



Science of Cooking – BCBT100
Spring 2014 - Bodwin



Common Sugars

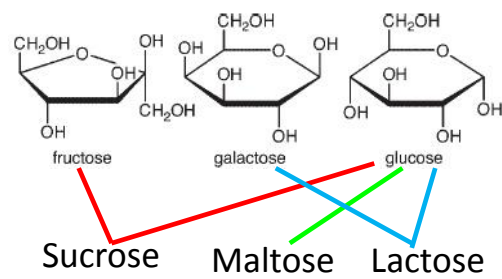


Image: <http://en.wikipedia.org/wiki/File:Glycolysis2.png>



Science of Cooking – BCBT100
Spring 2014 - Bodwin



Common Sugars

Relative sweetness of sugars and sweeteners

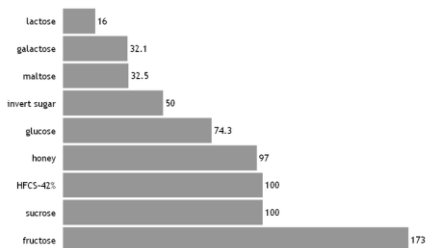


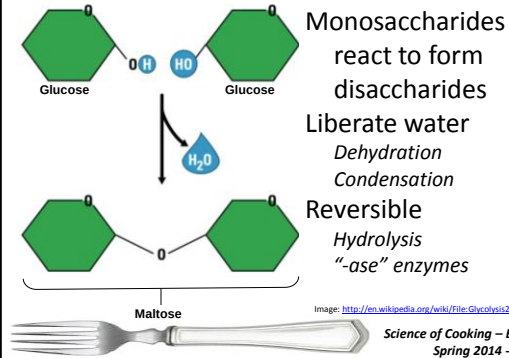
Image: <http://en.wikipedia.org/wiki/File:RelativeSweetness.png>



Science of Cooking – BCBT100
Spring 2014 - Bodwin

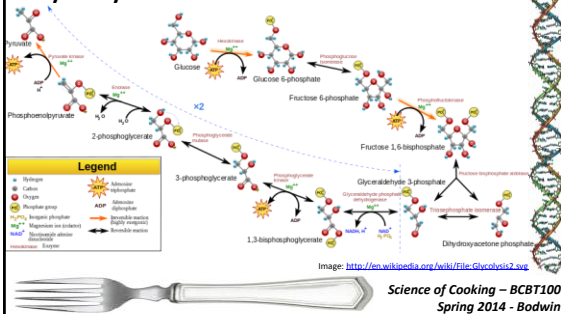


Disaccharides

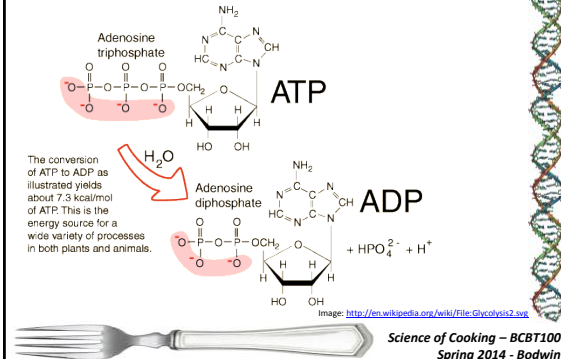


Sugar Metabolism

Glycolysis

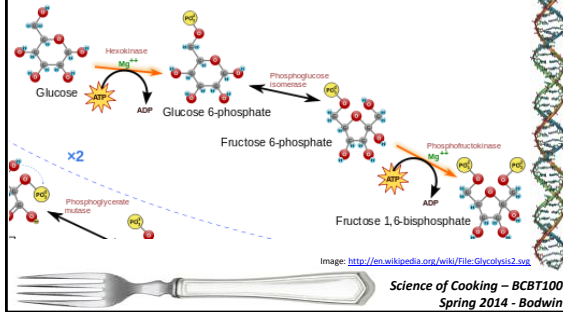


ATP $\rightleftharpoons$ ADP $\rightleftharpoons$ ATP



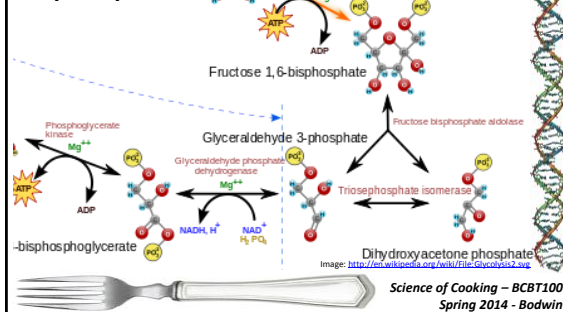
Sugar Metabolism

Glycolysis

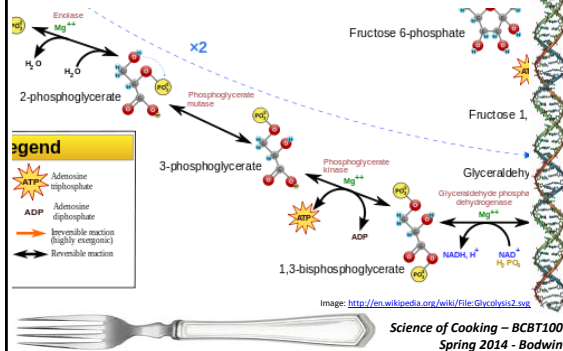


Sugar Metabolism

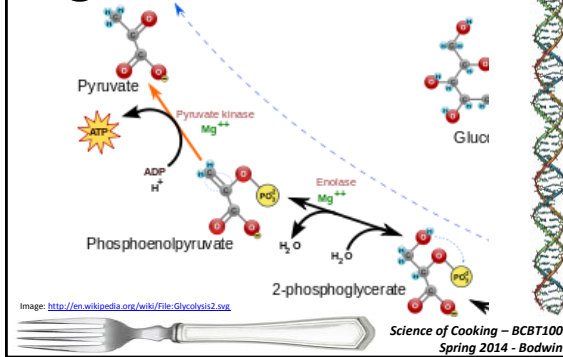
Glycolysis



Sugar Metabolism

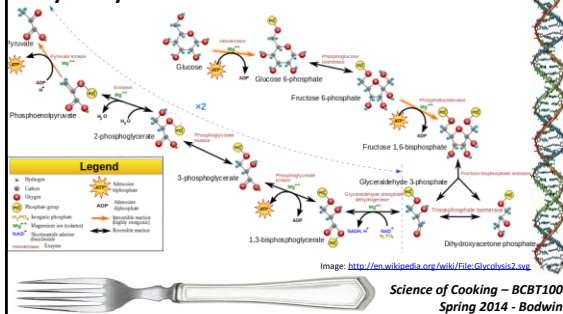


Sugar Metabolism



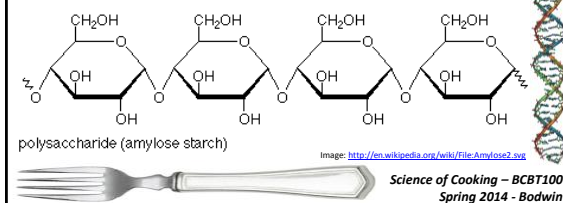
Sugar Metabolism

Glycolysis



Polysaccharides

Storage and structure
Starch, Glycogen, Cellulose
Sugar polymers



Starch

Glucose polymers
Energy storage in plants
Potatoes, rice, grains

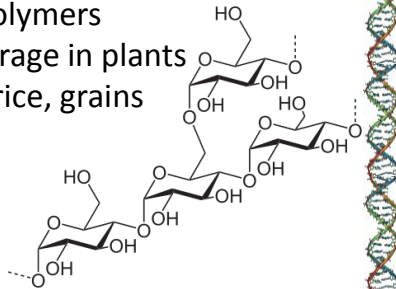


Image: http://en.wikipedia.org/wiki/File:Amylopectin_Sessel.jpg

Science of Cooking – BCBT100
Spring 2014 - Bodwin

Starch - Structure

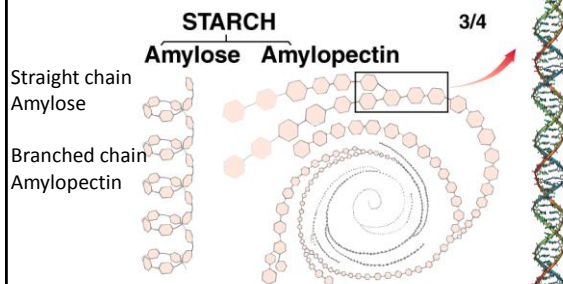
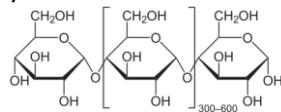


Image: <http://en.wikipedia.org/wiki/File:Amylose2.jpg>

Science of Cooking – BCBT100
Spring 2014 - Bodwin

Starch – In foods

Thickener – binds a LOT of water
Provides energy - amylase



Industrially:

Dextrose = glucose derived from
hydrolyzed starch

HFCS – dextrose treated with glucose
isomerase

Image: <http://en.wikipedia.org/wiki/File:Amylose2.jpg>

Science of Cooking – BCBT100
Spring 2014 - Bodwin

Glycogen – “animal starch”

Highly branched glucose polymer

Energy storage

GLYCOGEN

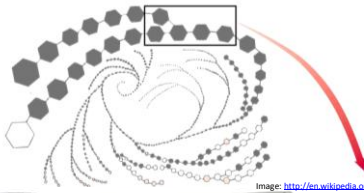
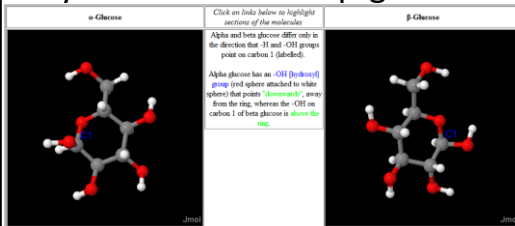


Image: <http://en.wikipedia.org/wiki/File:Amylose2.svg>

Science of Cooking – BCBT100
Spring 2014 - Bodwin

Cellulose

Polymers made from β -glucose



Side-by-side animations:

<http://www.biotopics.co.uk/JmolApplet/alphabetaajglucose2.html>

Science of Cooking – BCBT100
Spring 2014 - Bodwin

Cellulose

Enzymes that break amylose

can't break cellulose

Rigid, tough *fibers* that make

plant cell walls and stalks

Cross-linking



Science of Cooking – BCBT100
Spring 2014 - Bodwin

Cellulose - Dietary

Insoluble Fiber

Highly modified cellulose, up to ~1/2 the mass of a plant
 Binds water, "feel full"
 Draws water into gut
 Fruits, vegetables, whole grains



Science of Cooking – BCBT100
 Spring 2014 - Bodwin

Cellulose - Dietary

Soluble Fiber

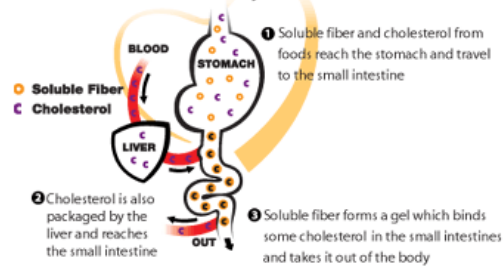
Highly modified cellulose
 Forms gel with high water content
 Water-soluble substances absorbed by gel – "intestine sweeper"



Science of Cooking – BCBT100
 Spring 2014 - Bodwin

Cellulose - Dietary

How Soluble Fiber May Lower Cholesterol



Science of Cooking – BCBT100
 Spring 2014 - Bodwin

Cellulose — Food source?

Cellulosic fiber is indigestible

Most animals lack enzymes to break down cellulose

Ruminants have bacteria in the gut that {partially} digest cellulose to glucose



Image: <http://www.publicdomainpictures.net/view-image.php?image=6278&picture=black-cow> <http://www.cem.ncsu.edu/vic/fda/cf/rhm/>



*Science of Cooking – BCBT100
Spring 2014 - Bodwin*

Interactions

Fats and water

Amphiphiles

Micelles

Emulsifiers



*Science of Cooking – BCBT100
Spring 2014 - Bodwin*
